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Crystallography of the Silver-Rich Rare-Earth-Silver Intermetallic Compounds

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The crystal structures of the silver-rich compounds in the rare-earth-silver systems were determined by means of X-ray powder diffraction methods. Metallographic methods were used to confirm the stoichiometries. The low temperature form of LaAg_5 crystallizes in a 'substituted' MgZn_2 ($C14$) type Laves-phase structure. The high temperature polymorph of LaAg_5 and the phases CeAg_5 , PrAg_5 and YbAg_5 were found to be isotypic but the crystal structure was not established. EuAg_5 crystallizes in the CaCu_5 ($D2_a$) type structure; LuAg_4 and ScAg_4 of the MoNi_4 ($D1_a$) type were found. $R_{14}\text{Ag}_{51}$ phases of the $\text{Gd}_{14}\text{Ag}_{51}$ type exist for $R = \text{Y, La, Ce, Pr, Nd, Sm, Gd, Tb, Dy, Ho}$ and Er . Yb_2Ag_7 crystallizes in the Ca_2Ag_7 type structure. TmAg_3 was found to exist in both the AuCu_3 ($L1_2$) type and the TiCu_3 ($D0_a$) type structures. The crystallographic data for these phases are given and their structural relationships are discussed.

Introduction

During an investigation of the solubility limits of the rare earths in silver, it became necessary to determine the stoichiometry of the first silver-rich compound in the rare-earth-silver systems (Gschneidner, McMasters, Alexander & Venteicher, 1970). Since most of these compounds are new, their structural properties and crystallographic data were determined and are the subject of this paper.

All of the rare earths except europium form the RAg compound with the cubic CsCl ($B2$) type structure. These phases have been studied by several investigators and most of their results are given in Pearson (1967) and McMasters & Gschneidner (1964). For YbAg and YAg the data are given by Moriarty, Humphreys, Gordon & Baenziger (1966). Iandelli & Palenzona (1969) have shown that YbAg crystallizes in both the CsCl and FeB type structures. Köster & Meixner (1965) reported that EuAg exists but the structure was not established.

For $R = \text{La, Ce, Pr, Nd, Eu}$ and Yb the phase RAg_2 crystallizes in the orthorhombic CeCu_2 (KHg_2) type structure according to Iandelli & Palenzona (1968) and Köster & Meixner (1965) in the case of EuAg_2 . For $R = \text{Gd, Tb, Dy, Ho, Er, Tm, Lu, Sc}$ and Y the

RAg_2 phase of the tetragonal MoSi_2 ($C11_b$) type structure has been reported by several investigators: Baenziger & Moriarty (1961), Moriarty, Gordon & Humphreys (1965), Dwight, Downey & Conner (1967), Kusma & Laube (1965), Ferro, Capelli, Borsese & Delfino (1967), Iandelli & Palenzona (1968), and Steeb, Godel & Lohr (1968). Different structures have been reported for SmAg_2 by Iandelli & Palenzona (1968), for YbAg_2 by Moriarty, Humphreys, Gordon & Baenziger (1966), and for YAg_2 by Gebhardt, von Erdberg & Lütz (1964).

The structure data for the RAg_3 phases reported by Steeb, Godel & Lohr (1968) and Donalato & Steeb (1969) will be discussed later in connection with the results of our investigation.

Lattice parameters and structure types for Yb_5Ag_3 (Cr_5B_3 type) and Yb_3Ag_2 (U_3Si_2 type) were determined by Iandelli & Palenzona (1969). Palenzona (1967) and Köster & Meixner (1965) reported EuAg_5 as being of the hexagonal CaCu_5 ($D2_a$) type structure. The latter also reported that the phases EuAg_4 and Eu_5Ag_3 exist but did not determine the structures.

Experimental procedures

The purity of the metals used in this investigation and alloy preparation procedures are described by Gschnei-

ner *et al.* (1970). Samples, weighing 5–8 g, of the stoichiometries RAg_6 , RAg_5 , RAg_4 , R_2Ag_7 , and RAg_3 were prepared for most of the R -Ag systems by arc-melting techniques. For $R=Eu$ and Yb the samples were prepared by induction heating the constituents in sealed tantalum crucibles. Weight losses were found to be negligible and heat treatments were carried out with the alloys sealed in tantalum crucibles. All samples were examined by metallographic and X-ray powder techniques. The metallographic results were used to confirm the compound stoichiometries and are given in detail by Gschneidner *et al.* (1970). The powder specimens were prepared under an inert gas atmosphere and were sealed in Pyrex capillaries for use in the X-ray analysis.

A computer program written by Yvon, Jeitschko & Parthé (1969) was used to interpret the powder patterns. The extrapolated lattice parameters were determined by a least-squares fit of the back-reflection powder pattern data and the Nelson–Riley function by means of a computer program written by Vogel & Kempter (1961). The calculated $\sin^2\theta$ values listed in the text are based on the extrapolated lattice parameters for the compounds and are therefore less than the observed $\sin^2\theta$ values since the latter contain the sample absorption error. The magnitude of the differences ($\sin^2\theta_c - \sin^2\theta_o$) is reflected in the slope of the straight line obtained by the extrapolation program. The accuracy of the lattice parameters (reported σ_s) corresponds to the standard deviation calculated for the observed reflection data in reference to the fitted straight line. Both $Cr K\alpha$, $\lambda = 2.29092 \text{ \AA}$ ($\lambda_1 = 2.28962$ and $\lambda_2 = 2.29351 \text{ \AA}$) and $Cu K\alpha$, $\lambda = 1.54178 \text{ \AA}$ ($\lambda_1 = 1.54050$ and $\lambda_2 = 1.54434 \text{ \AA}$) were used in conjunction with a 114.6 mm Debye–Scherrer camera.

Results

Lattice parameters and structure types of the new rare-earth–silver compounds studied in this investigation are listed in Table 1.

RAg_5 , 83.3 at.% Ag

The compounds $LaAg_5$, $CeAg_5$, $PrAg_5$, $YbAg_5$, and $EuAg_5$ were found to exist. The powder pattern data for the low temperature modification of $LaAg_5$ are given in Table 2. The sample which yielded this pattern had been heat treated at 500°C for 2 weeks. The powder pattern was indexed on the basis of the $MgZn_2$ (C14) type Laves-phase structure and the agreement between the observed and calculated intensity and $\sin^2\theta$ data is excellent. This agreement is retained when the AB_2 stoichiometry is shifted to AB_5 by randomly substituting two Ag atoms for two La atoms per unit cell. It was therefore concluded that $LaAg_5$ crystallizes in a ‘substituted’ $MgZn_2$ (C14) type structure. The calculated pattern based on the $LuMn_5$ (prototype to C14) structure reported by Wang & Gilfrich (1966) accounts for the more intense reflections of the $LaAg_5$ pattern; however, several discrepancies between the calculated and observed less intense reflections occur. For this reason the $LuMn_5$ (C14) prototype structure was eliminated as a possible solution for this structure.

The powder pattern of the $LaAg_5$ sample that had been heat treated at 780°C for 2 days contained several more reflections than that of the low temperature form but was not extremely complex. Attempts to index the pattern were unsuccessful, but it was determined that the structure is closely related to the low temperature form. This became apparent during the comparison of the observed $\sin^2\theta$ values with those calculated for the known Laves and Laves-related structures. The b and c type variations of the $CaCu_5$ type structure reported by Bronger (1967) for some RPT_5 phases could not be compared since the structural data are incomplete.

The $CeAg_5$, $PrAg_5$, and $YbAg_5$ phases are isostructural with the high temperature $LaAg_5$ phase as evidenced by the similarity in their powder patterns. Various heat treatments failed to produce these three RAg_5 compounds in the low temperature $LaAg_5$ crystalline form. The powder patterns of $PrAg_5$ and $YbAg_5$ are nearly identical and they probably have nearly the

Table 1. Structure data for some rare-earth–silver binary intermetallic compounds

Compound	Crystal system	Space group	Structure type	Lattice constants		
				a_0	b_0	c_0
$LaAg_5$ (low temp.)	hexagonal	$P6_3/mmc$	‘substituted’ $MgZn_2$ (C14)	$5.5690 \pm 0.0003 \text{ \AA}$	—	$9.0775 \pm 0.0005 \text{ \AA}$
$EuAg_5$	hexagonal	$P6/mmm$	$CaCu_5$ (D2a)	5.6201 ± 0.0008	—	4.6439 ± 0.0008
$LuAg_4$	tetragonal	$I4/m$	$MoNi_4$ (D1a)	6.6696 ± 0.0002	—	4.1581 ± 0.0002
$ScAg_4$	tetragonal	$I4/m$	$MoNi_4$ (D1a)	6.5740 ± 0.0002	—	4.0686 ± 0.0008
$La_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.955 ± 0.005	—	9.525 ± 0.007
$Ce_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.883 ± 0.005	—	9.455 ± 0.008
$Pr_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.846 ± 0.003	—	9.446 ± 0.005
$Nd_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.814 ± 0.002	—	9.432 ± 0.006
$Sm_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.750 ± 0.003	—	9.381 ± 0.002
$Gd_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.681 ± 0.003	—	9.289 ± 0.004
$Tb_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.650 ± 0.005	—	9.280 ± 0.003
$Dy_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.635 ± 0.003	—	9.271 ± 0.002
$Ho_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.609 ± 0.006	—	9.257 ± 0.002
$Er_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.596 ± 0.002	—	9.236 ± 0.001
$Y_{14}Ag_{51}$	hexagonal	$P6/m$	$Gd_{14}Ag_{51}$	12.637 ± 0.003	—	9.300 ± 0.002
Yb_2Ag_7	hexagonal	$P6_322$	Ca_2Ag_7	5.463 ± 0.001	—	14.084 ± 0.005
$TmAg_3$	cubic	$Pm\bar{3}m$	$AuCu_3$ (L1 ₂)	4.2117 ± 0.0005	—	—
$TmAg_3$	orthorhombic	$Pmnm$	$TiCu_3$ (D0 _a)	6.075 ± 0.005	4.948 ± 0.002	5.163 ± 0.003

Table 2. Calculated and observed powder pattern data for LaAg₅, low temperature form

Cu K α radiation. Atomic scattering factors for atoms in the 4(*f*) ($\frac{1}{3}, \frac{1}{3}, z$; etc.) positions are an average of f_{La} and f_{Ag} .

LaAg ₅ (C14) MgZn ₂ type						
sin ² θ_o	<i>I</i> _o	<i>h</i>	<i>k</i>	<i>l</i>	<i>I</i> _c	
0-0780	<i>ms</i>	1	1	0	0-0767	307
0-0919	<i>s</i>	1	0	3	0-0905	703
0-1035	<i>w</i>	2	0	0	0-1022	107
0-1070	<i>vs</i>	1	1	2	0-1055	1000
0-1108	<i>vs</i>	2	0	1	0-1094	816
0-1169	<i>mw</i>	0	0	4	0-1154	184
0-1327	<i>mw</i>	2	0	2	0-1310	162
0-1430	<i>mw</i>	1	0	4	0-1409	100
0-2327	<i>w</i>	3	0	0	0-2299	57
0-2454	<i>ms</i>	2	1	3	0-2438	274
0-2606	<i>ms</i>	3	0	2	0-2588	244
0-2843	<i>ms</i>	2	0	5	0-2825	313
0-2962	<i>w</i>	2	1	4	0-2942	80
0-3086	<i>ms</i>	2	2	0	0-3066	274
0-3634	<i>m</i>	2	0	6	0-3618	116
0-3990	<i>m</i>	3	1	3	0-3970	133
0-4174	<i>m</i>	4	0	1	0-4160	95
0-4235	<i>m</i>	2	2	4	0-4220	136
0-4399	<i>mw</i>	{ 2 1 6			0-4385	21
		{ 4 0 2			0-4376	23
0-4494	<i>w</i>	3	1	4	0-4475	31
0-4577	<i>w</i>	2	0	7	0-4556	25
0-4882	<i>w</i>	1	0	8	0-4871	45
0-5392	<i>w</i>	{ 4 1 0			0-5365	32
		{ 1 1 8			0-5382	32
0-5527*	<i>m</i>	3	2	3	0-5503	99

Table 2 (cont.)

sin ² θ_o	<i>I</i> _o	<i>h</i>	<i>k</i>	<i>l</i>	sin ² θ_c	<i>I</i> _c
0-5670	<i>s</i>	{ 4 1 2			0-5654	156
		{ 2 2 6			0-5662	76
0-5908	<i>ms</i>	4	0	5	0-5891	114
0-6418	<i>m</i>	2	1	8	0-6404	88
0-6704	<i>mw</i>	4	0	6	0-6684	61
0-7048	<i>w</i>	5	0	3	0-7036	35
0-7204	<i>m</i>	3	3	2	0-7187	75
0-7240	<i>m</i>	4	2	1	0-7226	129
0-7459	<i>mw</i>	{ 4 2 2			0-7442	34
		{ 3 2 6			0-7451	30
0-7437	<i>w</i>	2	1	9	0-7630	27
0-7951	<i>m</i>	3	1	8	0-7937	86
0-8244	<i>mw</i>	2	0	10	0-8234	77
0-8385	<i>w</i>	3	2	7	0-8388	49
0-8578	<i>ms</i>	5	1	3	0-8569	102
0-8964	<i>s</i>	4	2	5	0-8957	347
0-9078	<i>mw</i>	5	1	4	0-9074	50
0-9170	<i>mw</i>	3	1	9	0-9163	51
0-9205	<i>ms</i>	6	0	0	0-9198	200
0-9474	<i>mw</i>	3	2	8	0-9470	140
0-9752	<i>s</i>	{ 2 0 11			0-9748	357
		{ 4 2 6			0-9750	401

* Observed value consists of $K\alpha_1$ and $K\alpha_2$ converted to $K\alpha$ (mean) data in order to simplify the listing. This holds for values from asterisk to end of Table.

same lattice parameter values, which suggests that the Yb in this compound is divalent.

It is conceivable that the stoichiometry of these

Table 3. Calculated and observed patterns of the tetragonal MoNi₄ (*D*_{1a}) type structures of LuAg₄ and ScAg₄

Cu K α (mean) radiation.

LuAg ₄ $a_0=6.6696, c_0=4.1581 \text{ \AA}$				ScAg ₄ $a_0=6.5740, c_0=4.0686 \text{ \AA}$				
sin ² θ_o	<i>I</i> _o	sin ² θ_c	<i>I</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>I</i> _c	
—	—	0-0267	26	1	1	0	—	
0-0487	<i>vw</i>	0-0477	27	1	0	1	0-0275	49
0-0527	<i>vw</i>	0-0534	12	2	0	0	0-0504	49
0-1015	<i>vs</i>	0-1012	1000	1	2	1	0-0553	21
0-1334	<i>s</i>	0-1336	331	3	1	0	0-1057	1000
0-1379	<i>s</i>	0-1375	158	0	0	2	0-1385	331
—	—	0-1546	6	3	0	1	0-1452	154
—	—	0-1642	5	1	1	2	0-1603	10
0-1915	<i>vw</i>	0-1909	4	2	0	2	0-1733	9
0-2082	<i>vw</i>	0-2080	7	3	2	1	0-1733	9
—	—	0-2444	3	2	2	2	0-2005	7
0-2664	<i>ms</i>	0-2672	110	2	4	0	0-2157	13
0-2707	<i>s</i>	0-2711	215	3	1	2	0-2549	5
0-3681	<i>s</i>	0-3684	256	5	0	1	0-2769	108
0-3752	<i>s</i>	0-3761	124	1	2	3	0-2832	209
0-4037	<i>s</i>	0-4047	111	2	4	2	0-3816	248
—	—	0-4218	2	2	5	1	0-3935	118
—	—	0-4830	2	2	3	3	0-4205	106
0-5380	<i>w</i>	0-5344	39	6	2	0	0-4365	5
0-5521	<i>w</i>	0-5499	19	0	0	4	0-5022	4
—	—	0-5917	1	5	3	2	0-5510	38
0-6384*	<i>m</i>	0-6355	71	3	6	1	0-5743	18
0-6453	<i>ms</i>	0-6433	142	5	0	3	0-6121	4
0-6705	<i>m</i>	0-6680	72	5	5	0	0-6560	72
0-6743	<i>m</i>	0-6719	71	6	2	2	0-6684	144
0-6860	<i>m</i>	0-6835	71	3	1	4	0-6900	72
0-8071	<i>s</i>	0-8055	166	5	5	2	0-6946	73
0-8184	<i>s</i>	0-8171	86	2	4	4	0-7130	75
0-9035	<i>s</i>	0-9027	234	7	4	1	0-8321	176
0-9114	<i>s</i>	0-9105	122	3	6	3	0-8500	93
0-9267	<i>s</i>	0-9260	134	1	2	5	0-9300	274
							0-9423	151
							0-9664	199

* See footnote for Table 2.

compounds is not exactly RAg_5 since metallographic evidence was used to establish it. A complete crystal structure determination should establish the true stoichiometry for this phase.

$EuAg_5$ crystallizes in the $CaCu_5$ ($D2_d$) type Laves-related structure. The lattice parameters listed in Table 1 are in reasonable agreement with reported values.

RAg_4 , 80.0 at.% Ag

The silver-richest compound in the Lu-Ag and Sc-Ag systems was found to be of the RAg_4 stoichiometry. Both $LuAg_4$ and $ScAg_4$ crystallize in the $MoNi_4$ ($D1_a$) type structure. Since these are new compounds, their powder pattern data are given in Table 3 for comparative purposes. The agreement between the observed and calculated data is good. The atomic positional parameters established by Harker (1944) for the $8(h)$ set ($x=0.200$, $y=0.400$, $z=0.0$) were used in these calculations.

$R_{14}Ag_{51}$, 78.5 at.% Ag

The $R_{14}Ag_{51}$ phase is the silver-richest compound in the Nd-, Sm-, Gd-, Tb-, Dy-, Y-, Ho-, and Er-Ag systems. This phase crystallizes in the new $Gd_{14}Ag_{51}$ type structure. Single-crystal X-ray diffraction methods have been used by Bailey & Kline (1970) to complete the structure determination of this compound. The atom position data listed in Table 4 were supplied by them and were used to calculate the powder patterns of these $R_{14}Ag_{51}$ phases. Their observed and calculated intensities yield an R index of 4.5. Metallographically the alloys in these systems are two phase at 75 at.% Ag and single phase between 77.8 and 80.0 at.% Ag, in agreement with the structure determination of Bailey & Kline (1970). The powder patterns of these phases could be properly indexed on the basis of these structural data and the measured values for $Tb_{14}Ag_{51}$ listed in Table 5 are in good agreement with the calculated data.

Table 4. Crystallographic data for the $Gd_{14}Ag_{51}$ type structure

The refined data will be given by Bailey & Kline (1970) in their report on the complete structure determination.

$Gd_{14}Ag_{51}$ stoichiometry with disordered set.*
Hexagonal - space group $P6/m-C_{6h}^1$.

	Wyckoff notation	x	y	z
Ag	2(c)	0.333	0.667	0.0
Ag	4(h)	0.333	0.667	0.298
Ag	6(k)	0.239	0.059	0.5
Ag	12(l)	0.115	0.494	0.153
Ag	12(l)	0.266	0.192	0.237
Ag	12(l)	0.440	0.104	0.331
Gd	2(e)	0.0	0.0	0.307
Gd	6(j)	0.389	0.114	0.0
Gd	6(k)	0.141	0.469	0.5
3Ag*	6(j)	0.112	0.135	0.0

Table 5. Calculated and observed powder patterns for $Tb_{14}Ag_{51}$ with the $Gd_{14}Ag_{51}$ type structure

Cr $K\alpha$ (mean) radiation.						
$\sin^2 \theta_o$	I_o	h	k	l	$\sin^2 \theta_c$	I_c
0.1440	<i>vw</i>	3	1	0	0.1421	72
		2	2	1	0.1464	53
0.1606	<i>vw</i>	3	0	2	0.1593	25
0.1769	<i>w</i>	4	0	0	0.1749	87
0.1939	<i>w</i>	{ 4	0	1	0.1902	34
		{ 2	2	2	0.1921	56
0.2050	<i>s</i>	{ 1	3	2	0.2031	589
		{ 3	1	2	0.2031	113
0.2096	<i>m</i>	2	3	0	0.2077	84
0.2150	<i>vs</i>	{ 2	1	3	0.2137	1000
		{ 1	2	3	0.2137	294
0.2251	<i>m</i>	{ 3	2	1	0.2230	48
		{ 2	3	1	0.2230	94
0.2317	<i>vs</i>	{ 4	1	0	0.2296	118
		{ 1	4	0	0.2296	827
0.2369	<i>m</i>	3	0	3	0.2355	269
0.2470	<i>s</i>	{ 0	0	4	0.2438	111
		{ 4	1	1	0.2448	349
		{ 1	4	1	0.2448	41
0.2557	<i>w</i>	1	0	4	0.2547	81
0.2711	<i>m</i>	{ 2	2	3	0.2683	67
		{ 2	3	2	0.2687	217
0.2750	<i>w</i>	5	0	0	0.2733	85
0.2800	<i>m</i>	3	1	3	0.2792	160
0.2890	<i>mw</i>	2	0	4	0.2875	107
0.2931	<i>mw</i>	{ 1	4	2	0.2905	97
		{ 4	1	2	0.2905	28
0.3405	<i>mw</i>	{ 5	1	0	0.3389	21
		{ 1	5	0	0.3389	31
0.3459	<i>m</i>	3	2	3	0.3448	114
0.3685	<i>w</i>	{ 1	4	3	0.3667	14
		{ 4	2	2	0.3671	27
0.4067	<i>w</i>	4	3	0	0.4045	25
0.4438	<i>m</i>	4	2	3	0.4432	64
0.4669	<i>vw</i>	4	3	2	0.4654	25
0.4764	<i>vw</i>	1	4	4	0.4734	28
0.5427	<i>m</i>	4	3	3	0.5416	58
0.5484	<i>m</i>	0	0	6	0.5485	95
0.5652	<i>mw</i>	5	2	3	0.5635	63
0.6108	<i>s</i>	4	1	5	0.6105	207
0.6252	<i>w</i>	2	1	6	0.6250	45
0.6312	<i>s</i>	6	2	2	0.6294	198
0.6490	<i>m</i>	3	4	4	0.6483	138
0.6633	<i>ms</i>	4	4	3	0.6619	151
0.6706	<i>ms</i>	5	2	4	0.6701	120
0.6847	<i>mw</i>	{ 4	5	1	0.6821	54
		{ 7	1	2	0.6841	41
0.6910	<i>s</i>	{ 6	3	0	0.6887	85
		{ 3	6	0	0.6887	52
		{ 1	3	6	0.6906	89
		{ 3	1	6	0.6906	71
0.7064	<i>s</i>	{ 2	6	3	0.7056	133
		{ 6	2	3	0.7056	28
0.7144	<i>s</i>	{ 1	6	4	0.7139	105
		{ 8	0	1	0.7149	82
0.7231	<i>mw</i>	4	0	6	0.7234	76
0.7340	<i>mw</i>	7	2	0	0.7325	69
0.7623*	<i>s</i>	1	7	3	0.7603	183
0.7779	<i>s</i>	4	1	6	0.7781	240
0.7956	<i>m</i>	2	7	2	0.7934	95
0.8221	<i>m</i>	5	0	6	0.8218	60
0.8378	<i>w</i>	5	5	1	0.8352	81
0.8556	<i>w</i>	4	2	6	0.8546	66
0.8664	<i>s</i>	{ 3	7	0	0.8637	131
		{ 1	7	4	0.8669	40
0.8798	<i>mw</i>	3	7	1	0.8789	104
0.8866	<i>m</i>	{ 9	0	0	0.8855	28
		{ 5	1	6	0.8874	128

Table 5 (cont.)

$\sin^2 \theta_o$	I_o	h k l	$\sin^2 \theta_c$	I_c
0.9026	<i>mw</i>	{ 9 0 1 4 4 5	0.9008 0.9057	55 45
0.9202	<i>w</i>	8 2 0	0.9183	90
0.9260	<i>ms</i>	7 3 2	0.9246	436
0.9329	<i>ms</i>	3 6 4	0.9325	291
0.9511	<i>m</i>	2 6 5	0.9494	118
0.9535	<i>s</i>	{ 4 3 6 2 3 7	0.9530 0.9543	123 170
		{ 0 0 8 1 4 7	0.9751 0.9761	129 229
0.9770	<i>s</i>	{ 7 2 4	0.9762	447

* See footnote for Table 2.

Steeb, Godel & Lohr (1968) reported lattice parameters for some $R\text{Ag}_3$ ($R=\text{La}, \text{Ce}, \text{Sm}, \text{Gd}, \text{Dy}, \text{Ho}, \text{Y}$ and Er) compounds. For $R=\text{Pr}, \text{Nd}, \text{Tb}$ and Tm the $R\text{Ag}_3$ lattice parameters were reported by Donalato & Steeb (1969). Their lattice parameters are close to those of the $R_{14}\text{Ag}_{51}$ phases found in this study. It is believed that their $R\text{Ag}_3$ lattice parameters are actually

for the $R_{14}\text{Ag}_{51}$ stoichiometry with the possibility of slight solubility ranges existing for these phases.

Yb_2Ag_7 crystallizes in the hexagonal Ca_2Ag_7 type

Table 6. Calculated and observed powder patterns for Yb_2Ag_7 with the Ca_2Ag_7 type structure

Cu $K\alpha$ radiation.					
I_o	$\sin^2 \theta_o$	h k l	$\sin^2 \theta_c$	I_c	
<i>w</i>	0.0483	0 0 4	0.0479		
<i>m</i>	0.0539	1 0 3	0.0535		
<i>s</i>	0.0755	1 0 4	0.0745		
<i>m</i>	0.0804	1 1 0	0.0797		
<i>s</i>	0.0832	1 1 1	0.0827		
<i>m</i>	0.0921	1 1 2	0.0916		
<i>s</i>	0.1022	1 0 5	0.1015		
<i>vs</i>	0.1073	{ 2 0 0 1 1 3	0.1062 0.1066		
<i>m</i>	0.1186	2 0 2	0.1182		
<i>m</i>	0.1278	1 1 4	0.1276		
<i>m</i>	0.1328	2 0 3	0.1332		
<i>m</i>	0.1361	1 0 6	0.1344		
<i>m</i>	0.1880	1 1 6	0.1875		
<i>w</i>	0.1935	0 0 8	0.1917		
<i>m</i>	0.2150	2 0 6	0.2141		

Table 7. Calculated and observed powder patterns for TmAg_3 which exists in both the cubic AuCu_3 ($L1_2$) and the orthorhombic TiCu_3 ($D0_a$) type structures

I_c	AuCu_3 type			Cu $K\alpha$ (mean) radiation. TmAg_3 observed powder pattern 75 at.% Ag		TiCu_3 type		I_c
	$\sin^2 \theta_c$	h k l	$\sin^2 \theta_o$	I_o	h k l	$\sin^2 \theta_c$		
			0.0885	<i>m</i>	2 0 1	0.0867	330	
			0.0910	<i>w</i>	0 0 2	0.0892	67	
			0.0982	<i>m</i>	0 2 0	0.0971	445	
1000	0.1005	1 1 1	0.1024	<i>vs</i>				
			0.1072	<i>vw</i>	1 0 2	0.1053	10	
			0.1124	<i>s</i>	2 1 1	0.1110	1000	
			0.1154	<i>ms</i>	0 1 2	0.1134	607	
489	0.1340	2 0 0	0.1356	<i>s</i>				
			0.1855	<i>mw</i>	2 2 1	0.1838	213	
			0.2605	<i>w</i>	4 0 0	0.2576	97	
322	0.2680	2 2 0	0.2705	<i>vs</i>	2 0 3	0.2651	174	
			0.3076	<i>m</i>	{ 2 3 1 0 3 2	0.3052 0.3076	200 124	
			0.3565	<i>w</i>	{ 4 2 0 0 0 4	0.3547 0.3567	114 30	
			0.3641	<i>mw</i>	2 2 3	0.3621	209	
381	0.3685	3 1 1	0.3704	<i>vs</i>	4 1 2	0.3711	183	
111	0.4020	2 2 2	0.4039	<i>ms</i>				
			0.4458	<i>vw</i>	4 2 2	0.4439	22	
			0.4745	<i>vw</i>	2 4 1	0.4751	47	
58	0.5360	4 0 0	0.5375*	<i>m</i>				
213	0.6365	3 3 1	0.6390	<i>s</i>	4 1 4	0.6386	57	
214	0.6700	4 2 0	0.6721	<i>m</i>				
			0.6939	<i>mw</i>	2 5 1	0.6935	84	
			0.7022	<i>w</i>	6 2 1	0.6991	39	
			0.7137	<i>mw</i>	4 2 4	0.7114	67	
			0.7366	<i>w</i>	4 4 2	0.7352	17	
			0.7832	<i>mw</i>	6 0 3	0.7803	66	
253	0.8040	4 2 2	0.8055	<i>s</i>	0 0 6	0.8026	30	
			0.8223	<i>ms</i>	6 3 1	0.8204	104	
			0.8343	<i>m</i>	4 3 4	0.8328	74	
			0.8416	<i>m</i>	2 3 5	0.8402	160	
			0.8730	<i>w</i>	0 6 0	0.8738	46	
			0.8797	<i>s</i>	6 2 3	0.8774	178	
488	0.9046	3 3 3	0.9053	<i>s</i>	0 2 6	0.8997	89	
			0.9525	<i>ms</i>	4 5 2	0.9536	292	
			0.9615	<i>mw</i>	2 6 1	0.9605	119	

* See footnote for Table 2.

Table 8. *Intermetallic compound formation in the rare-earth-silver systems*

Stoichiometry and structure differences yielding eight sequences.

Sequence number	Rare earth	R_5Ag_3	R_3Ag_2	RAg	RAg_2	RAg_3	R_2Ag_7	$R_{14}Ag_{51}$	RAg_4	RAg_5
1	La, Ce, Pr			×	×*			×		×†
2	Nd			×	×*			×		
3	Sm			×	×†			×		
4	Eu	×†		×†	×*				×†	×
5	Gd, Tb, Dy, Ho, Er, Y			×	×‡			×		
6	Tm			×	×‡	×				
7	Yb	×	×	×	×*		×			×†
8	Lu, Sc			×	×‡				×	

* CeCu₂, KHg₂ type.
† Structure type unknown.
‡ MoSi₂ type.

structure reported in Pearson (1958). The atomic parameters needed to calculate the powder pattern intensities are not known so the observed and calculated $\sin^2\theta$ values listed for comparison in Table 6 are given as evidence for this structure.

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